

	UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education	ww. Firenepapers.com
CANDIDATE NAME		
CENTRE NUMBER	CANDIDATE NUMBER	
COMBINED S	SCIENCE	0653/03

Paper 3 (Extended)

Candidates answer on the Question Paper.

No Additional Materials are required.

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.

Write in dark blue or black pen.

You may use a soft pencil for any diagrams, graphs, tables or rough working.

Do not use staples, paper clips, highlighters, glue or correction fluid.

DO NOT WRITE IN ANY BARCODES.

Answer all questions. A copy of the Periodic Table is printed on page 16.

At the end of the examination, fasten all your work securely together. The number of marks is given in brackets [] at the end of each question or part question.

iner's Use

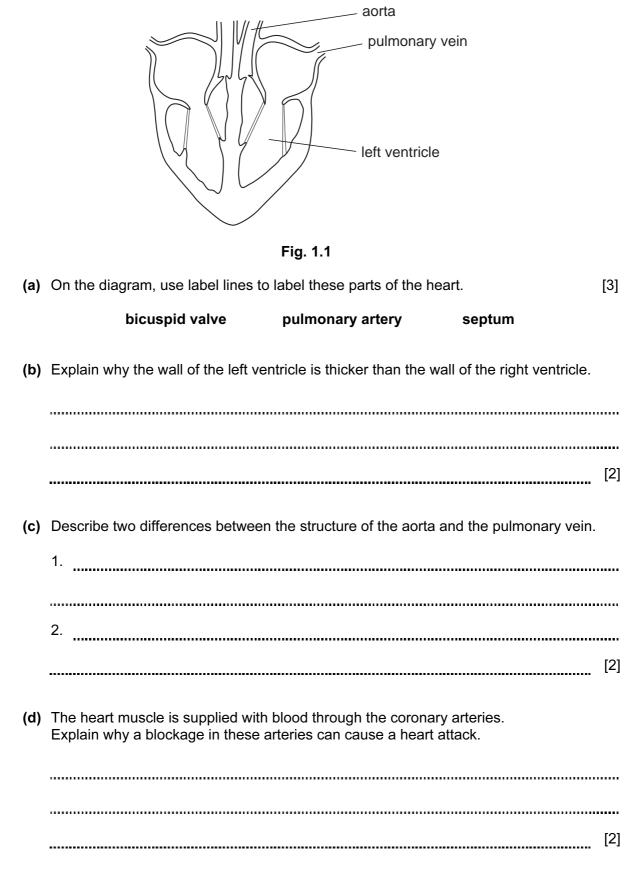
May/June 2007 1 hour 15 minutes

This document consists of 16 printed pages.

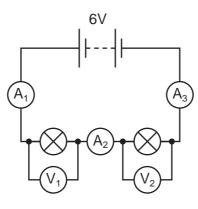


[Turn over

1 Fig. 1.1 shows a vertical section through a human heart.



2 (a) Fig. 2.1 shows a simple circuit containing two identical lamps.





Ammeter A_1 reads 0.15 A.

Write down the readings on

 ammeter A2,

 ammeter A3,

 voltmeter V1,

 voltmeter V2.

[2]

(b) (i) The electrical output from a power station is at 25000 V. The voltage is stepped up to 400000 V by a transformer. The number of turns on the primary coil is 20000.

Calculate the number of turns on the secondary coil.

State the formula that you use and show your working.

formula used

working

(ii) Explain why transformers require an a.c. input.
[3]

3 Fig. 3.1 shows a car in motion. The energy which is needed to make the car move comes from burning a mixture of air and fuel in the engine.

For Examiner's Use





(a) Air is a mixture of gases.

Describe **one** difference between a **mixture** of two gases and a **compound** formed from two gases.

[1]

- (b) Gasoline, a mixture of hydrocarbons, is a fuel used in car engines. When gasoline is burnt most of it undergoes complete combustion, but a small amount is incompletely combusted.
 - (i) Name **one** gaseous substance and **one** solid substance which are formed as the result of incomplete combustion.

gaseous substance solid substance [2]

(ii) Two chemical tests could be carried out on the mixture of exhaust gases to show that much of the gasoline fuel was undergoing **complete** combustion.

Describe these chemical tests.

 1.

 2.

 [4]

(c) The car battery contains sulphuric acid.
 (i) State the chemical formula of an alkali which would neutralise sulphuric acid to produce the salt, potassium sulphate.
 [1]
 (ii) Write a balanced equation involving ions which shows what happens when any acid is neutralised by any alkali.
 [2]

4 In Mexico, some areas of tropical rainforest have been cleared for growing cacao trees. Beans from cacao trees are used for making chocolate. The beans are seeds, and they develop from fertilised flowers.

For Examiner's Use

Bats are flying mammals that feed on insects, fruit or nectar. Many different bat species live in tropical rainforests.

Table 4.1 shows information about the numbers of plants and bats found in an undisturbed tropical rainforest and in a cacao plantation.

habitat	number of different species of plants	number of different species of bats	number of bat species found only in that habitat
in undisturbed rainforest	93	27	14
in cacao plantation	77	21	1

Table 4	4.1
---------	-----

(a) Explain how the data in Table 4.1 show that the rainforest has a higher species diversity than the cacao plantation.

[2]

(b) Using the data in Table 4.1, suggest **one** reason, other than species diversity, why leaving some areas of tropical rainforests undisturbed is important for the conservation of bats.

.....

......[1]

(c) Using the information provided, suggest how bats could help to increase the yield of beans from a cacao plantation.

[2]

(d) Farmers allow other plants to grow underneath the cacao trees.

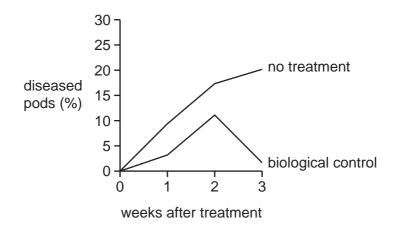
Explain how this could help to reduce soil erosion.

[2]

(e) Cacao trees are also grown in Africa. A fungus causes a disease called black pod, which can destroy up to 80% of the crop.

Farmers have found that the pesticides they have been using are no longer effective against this fungus. They have tried biological control instead, using a different fungus that attacks the black pod fungus.

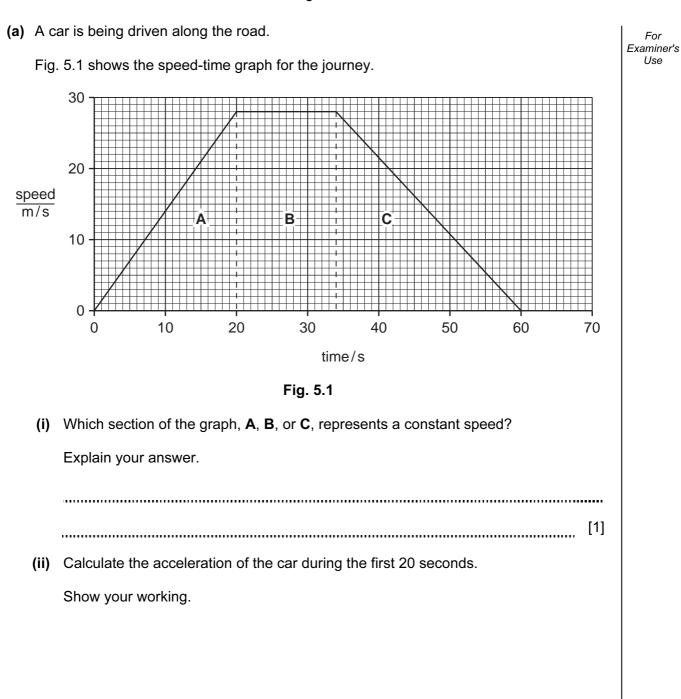
Fig. 4.1 shows the percentage of pods affected by black pod when no treatment was given and when the trees were treated with the biological control fungus.





(i) Describe the effect of the biological control fungus on black pod disease.

(ii) Suggest reasons for the changes in the number of diseased pods over the three week period when the biological control fungus was used.
 [2]



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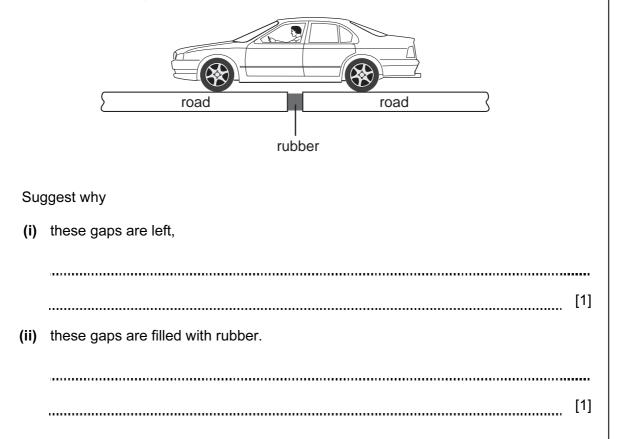
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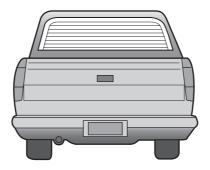
(b) The car travels over a long bridge. The bridge is made in sections, with gaps between each section. The gaps are filled with rubber.

For

Examiner's Use



(c) The heated rear windscreen of the car contains nine wires, connected in parallel, each with a resistance of 10 ohms.



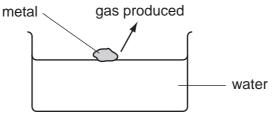
Is the combined resistance of all the wires more or less than 10 ohms?

Explain your answer.

[1]

6 (a) Fig. 6.1 shows a metal reacting in cold water.

A gas is produced very quickly during the reaction, and when this gas is tested it burns with a squeaky pop.





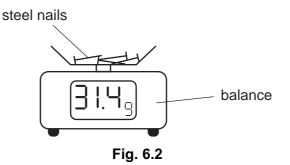
Suggest the name of a metal which would react like the one shown in Fig. 6.1.

Explain your answer.

(b) A student carried out an experiment into the rusting of steel nails. She used 31.0 g of new nails in her experiment.

After some days the nails had become rusty and the student re-weighed them.

Her result is shown in Fig. 6.2.



- (i) State the type of chemical reaction which takes place when steel rusts.
- (ii) Explain the increase in mass which the student found in her experiment.

[2]

.....

[1]

7 All metabolic reactions in animals and plants are catalysed by enzymes. Enzymes from plants usually have a lower optimum temperature than enzymes from humans.

Fig. 7.1 shows the rate of activity of a human enzyme at different temperatures.

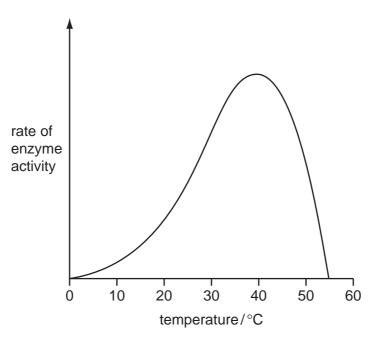


Fig. 7.1

(a) On Fig. 7.1, sketch a curve to show the rate of activity of a plant enzyme. [1]
(b) Explain the reasons for the shape of the curve for the human enzyme.
[4]
(c) Suggest why it is advantageous to a plant to have enzymes that have a lower optimum temperature than human enzymes. [1]

Gamm	a radiation and visible light are two regions of the electromagnetic spectrum.	For
(a) (i)		Examiner's Use
	[1]	
(ii)	All electromagnetic waves travel at the same speed in a vacuum.	
	State this speed.	
	[1]	
(iii)		
	[1]	
(b) Al	pha, beta and gamma are three types of radiation emitted during radioactive decay.	
(i)	State the meaning of the term radioactive decay.	
	[1]	
(ii)	Name a suitable detector for these three types of radiation.	
	[1]	
(iii)		
	alpha	
	beta	
	gamma	
	[3]	
(iv)	Describe how these types of radiation can be dangerous to the human body.	
	[2]	

8

9 The apparatus in Fig. 9.1 can be used to break down the compound lead bromide into its elements.



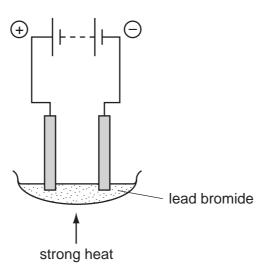


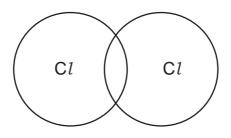
Fig. 9.1

(a) (i) Name the non-metallic element which is produced in this process.
[1]
(ii) Explain why the lead bromide shown in Fig. 9.1 has to be heated strongly in order for the process to work.
[2]
(b) Lead bromide has the chemical formula PbBr₂. Bromide ions are Br⁻.
(i) Deduce the charge on lead ions in lead bromide. Show how you obtained your answer. (ii) Deduce the total number of electrons in one bromide ion.

Explain how you obtained your answer.

number of electrons ______explanation ______[2]

- (c) A process similar to that in Fig. 9.1 is used in the chemical industry to produce the important element chlorine.
 - (i) Complete the bonding diagram below to show how the outer electrons are arranged in a chlorine molecule.



(ii) Chlorine reacts with the element silicon to form silicon chloride. In silicon chloride molecules, one silicon atom is bonded to four chlorine atoms.

Deduce a balanced symbolic equation for the reaction between silicon and chlorine.

[2]

[2]

						Gro	Group			≡	2	>	5		c
					⁺ Hydrogen					≡	2	>	>	>	2 Heium 4 O
				1						5 Born 1	6 Carbon	14 Nitrogen	16 Oxygen 8	9 Fluorine	20 Neon Neon
										27 A1 Aluminium 13	28 Silicon	31 Phosphorus 15	32 Sulphur 16	35.5 C1 Chlorine	40 Argon 18
51 52 55 V Cr Mn Vanadum Chromium Manganese 23 24 25	51 52 N Chromium 24 24 23 24 24 23 24 23 23 23 23 23 23 23 23 23 23 23 23 23	52 Tomium 2	55 Mn Manganese 25		56 Fe Iron	59 Co Cobatt 27	59 Nickel 28	64 Copper 29	65 Znc 30	70 Ga Gallium 31	73 Ge Germanium 32	75 AS Arsenic 33	79 Se Selenium 34	80 Bromine 35	84 Krypton 36
93 96 97 96 16 16 16 16 16 16 16 16 16 16 16 16 16	93 96 Nb No Molybdenum 42 42	4	Technetium 13		101 Ru Ruthenium	103 Rh Rhodium 45	106 Pd Palladium	108 Ag Silver	112 Cd Cadmium 48	115 In Indium	119 Sn 50	122 Sb Antimony 51	128 Te Tellurium 52	127 I Iodine 53	131 Xenon 54
181 184 186 Ta W Re Tantalum Tungsten 75	181 184 Ta V Intalum 75 75	184 V ungsten 75	186 Re Rhenium 75		190 OS Osmium 76	192 Ir Iridium 77	195 Pt Platinum 78	197 Au Gold 79	201 Hg ^{Mercury} 80	204 T 1 Thallium 81	207 Pb Lead	209 Bi Bismuth	Polonium 84	At Astatine 85	Rn ^{Radon}
140 141 144 Centum Praseodymium Neodymium 58 59 60	Cerium 259		144 Neodymium 30		Promethium 61	150 Sm Samarium 62	152 Eu 63	157 Gd Gadolinium 64	159 Tb ^{Terbium} 65	162 Dysprosium 66	165 HO Holmium 67	167 Er Erbium 68	169 Thulium 69	173 Yb Ytterbium 70	175 Lu Lutetium 71
232 238 Th Thorum Protactinum U 90 91 91 92	232 Th Protactinium 91	92	238 Uranium 32		Neptunium 93		Am Americium 95	C C C C C C C C C C C C C C C C C C C	BK Berkelium 97	Californium Californium	Einsteinium 99	Fm Fermium	Mendelevium 101	Nobelium 102	Lawrencium 103

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