

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

MARK SCHEME for the May/June 2012 question paper

for the guidance of teachers

0653 COMBINED SCIENCE

0653/23

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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	Page 2	2	Mark Scheme: Teachers' version	Syllabus	Paper		
			IGCSE – May/June 2012	0653	23		
1			e) distance / time ; = 12.5 (km / h) ;		[2]		
	(b) (i)	cher	nical ;		[1]		
	(ii)	(ii) heating engine / heating surroundings / light / sound ;					
	me	 (c) metal track expands in summer / hot weather ; metal can expand into gap ; prevents damage to tracks ; 					
	P. 6				[max 2]		
	(d) eth	(d) ethanol is renewable source / no sulfur dioxide produced ;					
	(e) (i)	 (e) (i) 5 (km/h); (accept 4 / reference to greater than 3 but equal to and less th (ii) 1200 (W); 		ual to and less than t	5) [1]		
	(ii)				[1]		
	(iii)	wind	speed variable / wind unreliable / owtte ;		[1]		
					[Total: 10]		
2	(a) (i)	nucl	eus ;		[1]		
	(ii)	18; evid	ence of neutrons = nucleon number minus proton n	umber ;	[2]		
	(iii)	hydr	rogen ;		[1]		
	(b) (i)	e.g.	typical metal property for X and corresponding non- X conductor Y insulator X malleable Y not malleable	metal for Y ;	[1]		
	(ii)	Y is	reactive / specific example, ${f Z}$ is unreactive ;		[1]		
	(c) (i)		rence to oxygen (not air) ; ch joins / reacts with carbon ;		[2]		
	(ii)	 soil (too) acidic ; lime, reacts with / neutralises acid / reduces acidity ; to increase fertility ; 					
				[max 2]			
					[Total: 10]		

	Page 3			Mark Scheme: Teachers' version	Syllabus	Paper
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3	(a)	only eats		al that eats plants ; s plants / does not eat meat ; e to getting energy from, its food / plants ;		[max 2]
	(b)	eat		; e in, more energy than they use ; carbohydrate / protein, converted to fat ;		[max 2]
	(c)	(i)	 (i) the greater the body mass, the greater the chance of survival ; idea that effect is greater at lower body masses / levels off at higher body masses ; use of figures ; 			
		use				[max 2]
		(ii)	insu	lator / poor conductor / reduces conduction ;		[1]
	(d)	carbon dioxide ; methane ;				[2]
	(e)	(i)	(mea	an) body mass is increasing ;		[1]
		 (ii) marmots have more time to feed (from spring onwards) / marmots lose less weight during hibernation (as winters are shorter); as spring arrives earlier, plants grow faster and as marmots are herbivores they have more plants to eat, therefore increasing mass; 		nots are herbivores	[max 1]	
					[Total: 11]	
4	(a)	(a) add magnesium to acid ;				
			asure	ng ; time for known volume of gas to collect / measure v rvals / measure how long reaction lasts ;	volume of gas at fixed	[3]
	(b)	(i)	temp	perature / surface area of magnesium ;		[1]
		(ii)		for (same volume) of gas to collect is shorter ; ause rate of reaction is greater ;		[2]
	(c)	(i)	atom	s charged, atom is neutral / proton and electron num n, but differ in the ion / the atom is reactive, the ion is tron shells full, while atom outer shell is not full ;		[1]
		(ii) №		MgCl ₂ ;		
		(iii) hydrogen ;				[1]
			-			[Total: 9]
						[]

	Page 4			Mark Scheme: Teachers' version	Syllabus	Paper	
				IGCSE – May/June 2012	0653	23	
5	(a)	(i) between 10 and 20 (Hz) to between 20 000 and 25 000 (Hz) ;				[1]	
		[2]					
	(b)	(i)	need	ds, air / particles / a medium ;		[1]	
		(ii) quieter / lower volume ;					
	((iii)	micr	rowaves ;		[1]	
	(c)			incidence labelled ; refraction labelled ;		[2] [Total: 8]	
6	(a)	labe	el to r	oot hair cell ;		[1]	
	(b)	[2]					
	(c)	(i)	xylei	m ;		[1]	
		(ii)	A in	central area of root ;		[1]	

(d) (i)

structure	plant cells	animal cells
cell membrane	\checkmark	1
cell wall	\checkmark	×
nucleus	\checkmark	\checkmark
vacuole containing sap	\checkmark	×

1 mark for any row correct ;;;;

[4]

[1]

(ii) no, as they do not get light / as they are underground ;

[Total: 10]

	Page 5				Syllabus	Paper
			IGCSE – May/Jur	e 2012	0653	23
7	(a)	 (i) ammeter in series with lamp ; voltmeter in parallel with lamp ; variable resistor in series with lamp and all else correct ; 				[3]
		(ii)	to change, voltage across / current	through, lamp / in ci	rcuit ;	[1]
	(b)	(i)	positive and negative ;			[1]
		(ii)	electrons ;			[1]
	(iii) m		metal / named metal / graphite ;			[1]
						[Total: 7]
8	(a)	(i)	nitrogen 78% oxygen 21% others 1 (all correct 2, two correct 1)	% ;;		[2]
		(ii)	carbon dioxide / water (vapour) / an		[1]	
	(b)	odic Table / cannot	be			
		<i>(</i>)	contain different types of atom / ele	ments (bonded) ;		[2]
		(ii)	covalent ; bonded elements are both non-met	als / compounds are	e gases ;	[2]
		(iii) pure water is neutral; rain water (generally) is (more) acidic (than pure water); rainwater during thunderstorm is the most acidic / owtte; likely to be caused by pitragen ovideo;				
		likely to be caused by nitrogen oxides ; which have reacted with the rain to form an acid ;			[max 2]	
						[Total: 9]
9	 (a) nutrition excretion respiration reproduction growth movement (1 mark for any two correct) ;; 			[max 2]		
		(Thank for any two concord),			[]	
	(b)		adrenaline ;			[1]
		 (ii) increases pulse rate ; increases blood pressure ; increases glucose in blood ; 				
	(c) in the blood / blood vessels ;					[1]
	-					[Total: 6]
						- •