

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

MARK SCHEME for the May/June 2012 question paper

for the guidance of teachers

0653 COMBINED SCIENCE

0653/63

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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P	Page 2		Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – May/June 2012	0653	63
1 (a)) (i)		C; / temperature/optimal for enzymes/owtte;		[2]
	(ii)	205,	217, 185 ;;		[2]
	(iii)	202 s (allov	s ; w 1 mark max in parts (i) and (ii) if times only given	in minutes)	[1]
(b)	fat <u>fatt</u>	is dige <u>y</u> acid	to sodium carbonate ; ested/broken down ; s neutralise the alkali ; phenolphthalein to change colour/neutralise ;		[max 2]
	uat	Joing P			
(c)	(c) to ensure contents/tubes reach the temperature/all tubes the same temp/body temp;				
(d)	 (d) EITHER <pre>repeat with boiled/heated/denatured lipase (demonstrates it is an enzyme); no change in pink colour/no reaction/very long time to change colour; OR</pre> 				
	repeat with different types of fat or named fat (demonstrates it breaks down fats) reaction works as before/owtte ;			ts) ; [max 2]	
					[Total: 10]
2 (a)) 13.	7;			[1]
(b)) (i)	exter	th (<i>1</i>) = 7.8 ; rnal diameter, (d _e) = 2.5 ; nal diameter, (d _i) = 1.8 ;		[3]
	(ii)	2.5 ² - = 3.0	– 1.8 ² ; (allow ecf) 01 ;		[2]
	(iii)	- (V)) = $3.14 \times 3.01 \times 7.8 \div 4 = ;$ (allow ecf)		
		(betv	veen) 18.1 and 18.5 ;		[2]
(c)	 (c) (formula used) density = mass/volume ; 0.74 ; (allow ecf from incorrect values, but not from incorrect formula) 				[2]
					[Total: 10]

	Page 3		Mark Scheme: Teachers' version	Syllabus	Paper
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3	3 (a) 20. <u>0</u> ; 47.5		.5 ; 43.5 ; (no tolerance)		[3]
	(b) (i)	point	correct and both labelled with units ; is correctly plotted ; <u>oth</u> curve through points ;		
		maxi	mum ;		[4]
	(iii)	from	graph (should be about 34 but accept 32) ;		[1]
	(iii)		titution 25 × 4.2 × ans (b)(iii) ; ectly worked out if use 34 = 3360 ;		[2]
					[Total: 10]

[1]

4 (a) (i) correct answers in column 3;

time after drinking coffee/min	number of beats in 30 s	number of beats per min		
0	36	72		
5	39	78		
10	42	84		
15	45	90		
20	45	90		
25	37	74		
30	36	72		

(ii)	suitable axes (scale and labels) ; plotting correct ; decent curve drawn ;	[3]
<i></i>		
(iii)	correct estimate from graph (about 17.5) ; (do not allow range)	[1]
(b) (i)	exercise causes heart rate to increase (therefore not a fair test);	[1]
(ii)	volume of coffee ;	
(")	concentration of coffee ;	
	(amount of/quantity of coffee – max 1)	[2]
(iii)	take readings more frequently (e.g. every 2 minutes) ;	
(111)	would see more clearly the peak in heart rate ;	
	more readings between 15 and 20 minutes ;	[max 2]
		[Total: 10]
		[Total: 10]

	Page 4		Mark Scheme: Teachers' version	Syllabus	Paper	
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5	(a) (i)	i) 9 (cm) ;			[1]	
	(ii)	 × 2 = 540 (m); (iii) allow any sensible idea, e.g. distracted/forgot/not concentrating/didn't hear correct sound owtte; (NOT just timing / experimental error) (iv) 1.76(5); (allow 1.76 or 1.77) (v) using their value from above ÷ their distance; answer; 				
	(iii)					
	(iv)					
	(v)					
		e.g. 540 ÷ 1.765 = 306			[2]	
	(vi)	 (vi) must comment on their value, e.g. accurate as values are close together inaccurate as values far apart ; (b) any two of the following: longitudinal wave ; (requires) molecules/particles ; closer together ; 				
	lon (red					
6		a) <u>lighted</u> splint ; pops/small explosion etc ;			[2]	
	(b) (i)	bubb	bles/gas/hydrogen floats Mg to surface/owtte ;		[1]	
	(ii)	(cop	oper) doesn't react with <u>acid</u> ;		[1]	
			um + copper produces hydrogen faster/steeper gra acts as a catalyst/hydrogen given off faster (if say st		[2]	
	(d) sor	some magnesium/solid remains ;		[1]		
	• •	e) sketch below others ; (and) reaches same level ;				
	(f) cor) connected to a syringe (labelled or graduations shown);		[1]		
					[Total: 10]	