MARK SCHEME for the October/November 2006 question paper

0653 COMBINED SCIENCE

0653/03

Paper 3 (Extended Theory), maximum raw mark 80

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This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

The grade thresholds for various grades are published in the report on the examination for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses.

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UNIVERSITY of CAMBRIDGE International Examinations

Page	2	Mark Scheme	Syllabus	Paper
		IGCSE - OCT/NOV 2006	0653	3
1 (a)	(i)	peat / wood / straw / biomass / rubbish / biogas / biodiesel	/ hydrogen;	
	(ii)	68% (40 + 25 + 3);		
(b)	(i)	to reduce energy losses; allow 'heat loss' ignore 'power	r loss'	
	(ii)	transformers use a.c./cannot use d.c./so voltage can be st	epped up (o	r
		down);		
		alternating current produces changing magnetic field;		
	(iii)	sine wave centred on 0V;		
		amplitude and wavelength approximately steady;		

[Total 7]

Page 3	Mark Scheme	Syllabus	Paper
	IGCSE - OCT/NOV 2006	0653	3

2 (a) A oviduct / Fallopian tube

B amniotic fluid **C** cervix **D** umbilical cord one mark for any two correct; (b) through the placenta; from its mother's blood; by diffusion; along umbilical cord; (C) forms bond between mother and baby; breast milk contains antibodies; avoids possibility of bacterial contamination; (not 'clean' or 'pure') at right temperature; changes composition as baby grows; (d) virus / HIV, passed from mother to baby;

crosses the placenta / passes from mother's blood to baby's blood; [2]

[Total 9]

[max 2]

[2]

[max 3]

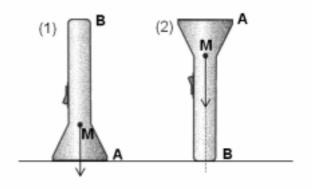
Page 4	Mark Scheme	Syllabus	Paper
	IGCSE - OCT/NOV 2006	0653	3

3 (a)	(i)	1 carbon dioxide;	
		2 hydrogen;	
		3 carbon dioxide;	[3]
	(ii)	HNO ₃ ; (not 'NHO ₃ ')	[1]
	(iii)	nitric acid;	[1]
(b)	(i)	experiment 6;	
		(in 6) time to collect same volume of gas was the shortest / greatest volume in	
		a given time;	[2]
	(ii)	(assume in experiment 6 [v.v. if describing experiment 5])	
		temperature could have been higher;	
		particles (moving faster) colliding more frequently (with the solid) / collisions	
		have more energy;	
		OR	
		acid concentration could have been higher;	
		more acid particles so greater collision frequency;	
		OR	
		greater surface area of solid;	
		so greater collision frequency (between solid and acid particles); [r	nax 2]

Page 5	Mark Scheme	Syllabus	Paper
	IGCSE - OCT/NOV 2006	0653	3

4 (a)	(i)	4.5V;	[1]
	(ii)	4.5V;	[1]

- (b) (i) A point at which the whole mass may be considered to act; [1]
 - (ii) diagram showing torch on end **B** and position of **M**;



centre of mass is closer to base;

base has larger area;

sensible tipping diagram /

[max 2]

[Total 5]

Page 6	Mark Scheme	Syllabus	Paper
	IGCSE - OCT/NOV 2006	0653	3

5	(a)	(i)	transported, in blood / to muscles;	
			respiration;	
			oxygen combined with glucose;	
			to form water and carbon dioxide;	[max 3]
		(ii)	20 kJ (dm ⁻³);	[1]
	(b)	(i)	0.3 (kJ per metre);	[1]
		(ii)	the longer the race, the less energy used (per metre);	
			run faster in shorter race / higher proportion of run is acceleration in shorter	
			race; there are other possible answers	[2]
	(c)	oxyg	en debt;	
		he h	ad been respiring anaerobically;	
		prod	ucing lactic acid;	
		whic	h must now be broken down using oxygen;	
		(allov	w marks for good description of carbon dioxide build-up and how this increase	<u>ə</u> s

(allow marks for good description of carbon dioxide build-up and how this increases breathing rate) [max 3]

[Total 10]

Paper	Syllabus	Mark Scheme	Page 7	
3	0653	IGCSE - OCT/NOV 2006	•	
		fractional distillation / fractionation;	(a) (i	6
		F;	(i	
		(1 x 2) + (1 x 16) / 18;	(b) (i	
	working;	(44 + 36) / 80g products / 32 000 is 2000 times 16 / oth	(i	
		2000 x 80 / 160 000 g ;		
		methane / oxygen;	(c) (i	
	•	idea that energy released is greater than energy absor	(i	
		ene molecules join (allow on diagram);	(d) e	
		a long chain (or diagram clearly implies this);	in	
[max		rence to how double bonds open to allow the linkage;	re	

Page 8	Mark Scheme	Syllabus	Paper
	IGCSE - OCT/NOV 2006	0653	3

7 (a)	(i)	series of straight lines + rays reflecting off walls of fibre,	
		at approx correct angles;	[1]
	(ii)	idea of interference / distortion / confusion in what is seen;	[1]
	(iii)	less interference / clearer information /	
		more messages can be sent at the same time /	
		signal needs boosting less often;	[1]
(b)	(i)	speed = distance / time = 1000 / 3;	
		= 333 m/s;	[2]
	(ii)	measure mass with, scales / balance;	
		measure volume;	
		displacement method for measuring volume described;	
		density = mass / volume;	[4]

Page	9	Mark Scheme	Syllabus	Paper	
		IGCSE - OCT/NOV 2006	0653	3	
3 (a)	(i)	phloem tubes are near surface (of stem);			[1]
	(ii)	phloem contains substances the plant has made;			
		sugar / sucrose / amino acids; not glucose,. not starch			
		xylem contains (mostly) water;		[ma	ax 2]
(b)	(i)	reach all parts of the plant (so kill all feeding insects);			
		only kill insects that eat the plant / do not kill beneficial inse	ects;		
		need to use less;			
		not washed away (by rain);		[ma	ax 2]
	(ii)	biological;			[1]
(c)	a gro	oup of cells;			
	simi	ar to each other / carrying out the same function;		[ma	ax 2]

[Total 8]

Page 10	Mark Scheme	Syllabus	Paper
	IGCSE - OCT/NOV 2006	0653	3

9 (a)

can be hammered into different shapes	Μ
poor conductor of heat	
is a gas at room temperature (20°C)	
good conductor of electricity	Μ
poor conductor of electricity	

		both required for one mark;	[1]
(b)		13;	[1]
(c)	(i)	aluminium ion electron config. 2.8;	
		charge 3+;	
		oxide ion electron config. 2.8;	
		charge 2-;	[4]
	(ii)	gains electrons / is discharged / becomes an (aluminium) atom;	
		(each ion gains) three electrons;	[2]
	(iii)	$2Al_2O_3 \rightarrow 4Al + 3O_2;$	[1]

Page 11		Mark Scheme	Syllabus	Paper	
		IGCSE - OCT/NOV 2006	0653	3	
10 (a)	partic	eles collide, more often / harder / faster;			
	with container walls;				[2]
(b)	spee	d is a scalar quantity / velocity is vector quantity;			
()	opeo				
	OR				
	velocity specifies direction but speed does not;			[1]	
(c)	alpha	will be absorbed by, air / skin, from outside;			
	dama	age cells within the body / mutation / damage DNA / cause	cancer;		
	not 'id	onises cells'			[2]

[Total 5]