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## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2007 question paper

## 0653 COMBINED SCIENCE

0653/02

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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	Page 2		Syllabus	Paper	
1	(a) leaf	IGCSE – October/November 2007	0653	<b>02</b> [1]	
	<b>` Q</b> to	cell membrane / vacuole membrane ; nucleus ; chloroplast ;		[3]	
	then rinse add	to <u>boiling</u> water ; to <u>hot</u> alcohol ; in water ; todine (solution) ; black colour produced;		[max 3]	
	(d) sexu antho	ers;		[3]	
2	(a) (i)	arrows going down/convection current with cold air dire	ection labeled;	[1]	
	(ii)	convection;		[1]	
	` '	(cold air) is denser/ has particles which are closer toge flows / drops to bottom of fridge;	ther;		
		displaces warmer air;		[2]	
	` '	$V \div I / \text{resistance} = \text{volts} \div \text{current/amps};$ 0 ÷ 0.04=6000 ( $\Omega$ );		[2]	
	polys by co	(c) aluminium reflects radiation back; polystyrene stops heat traveling through; by conduction /convection;			
	(salv is pro	nat the structure )	[3]		
3	(a) 4;			[1]	
	(b) (i)	speeds up the reaction;		[1]	
	(ii)	transition metals;		[1]	
	(c) (i)	covalent;		[1]	
	(ii)	O=O / fully correct dot and cross diagram;		[1]	
	(iii)	2. $H_2O_2 \rightarrow2. H_2O + O_2$ ;		[1]	

	Page 3		}	Mark Scheme	Syllabus	Paper
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4	(a)	n) respiration ;				
	(b)	(b) by decomposition (of dead organisms / bodies); decay organisms / detritivores / decomposers; example of decomposer e.g. bacteria/fungi; respire;				[max 2]
	(c) dead / once living organisms / plants / animals / bacteria; do not decay fully; in anaerobic / airless / absence of oxygen / waterlogged conditions; idea that they are compressed and buried; reference to long timescale				[max 2]	
	(d)	(i)	burn	ing fossil fuels / named fossil fuel / other fuels e.g. v	vood;	[1]
		(ii)		on dioxide concentration rose before humans were r implication that carbon dioxide levels high in the parity);		
		(iii)		al warming / temperature rise / <u>worsening</u> of greenheffect mentioned, e.g. sea level rise;	ouse effect;	[2]
5	(a)	(i)	_	ght / gravity; on/air resistance;		[2]
		(ii)	_	tht / gravity is greater than air resistance / $\mathbf{F_1}$ greater v ecf	than F <sub>2</sub> ;	[1]
	(b)	(b) (average) speed = distance/time; = 400 000/80= 5000 km/h;			[2]	
	(c)	(i)	there	e is no difference;		[1]
		(ii)	_	ght will be less on the moon / 900N on earth 150N or rent because gravity lower on moon;	n moon /	[1]
	(d)	sola	ar ene	ergy / sunlight;		[1]

Page 4	Mark Scheme	Syllabus	Paper
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6	(a)	rea	ction is exothermic / gives out heat (energy) / because of the heat released;	[1]
	(b)	(i)	the idea that there are two potassium atoms / ions for every one oxygen / two potassic particles are bonded to one oxygen/oxide;	um [1]
		(ii)	atom has same number of protons as electrons; positive ion has more protons than electrons;	[2]
	(c)		een) to purple / blue; etal oxides produce) alkaline solutions;	[2]
	(d)	(i)	KOH;	[1]
		(ii)	hydrogen; lighted splint; pops; allow ecf for correct test /result on incorrect gas	[3]
7	(a)	(i)	sub-Saharan Africa ;	[1]
		(ii)	the more HIV/AIDS, the more TB;	[1]
		(iii)	immune system cannot work properly / T cells do not work; unable to destroy TB <u>bacterium</u> ;	[2]
(b)			s oxygen taken in ; gen needed for energy release by <u>respiration</u> ;	[2]
	(c)	(i)	gonorrhoea; syphilis; (accept others e.g. chlamydia, genital warts, herpes)	[2]
		(ii)	use of condom / keeping to one partner / abstinence if a person has HIV / use of antibiotics; also allow the term, preservative, protection	[1]

Page 5	Mark Scheme	Syllabus	Paper
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8 (a) arrows in right direction;

ray of light from tooth to mirror and mirror to eye; approx correct angles;

[3]

(b) (i) a value in the range 10 to 20 Hz; a value in the range 20 000 to 25 000Hz;

[2]

(ii) number of waves produced/passing per second;

[1]

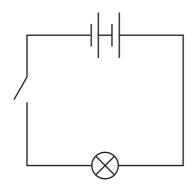
(iii) light/heat/thermal/nuclear/electrical/kinetic/potential/chemical;

[1]

(c) (i) one cell is back to front; ignore reference to blown bulb.

[1]

(ii)



[1]

Page 6		Mark Scheme	Syllabus	Paper
,		IGCSE – October/November 2007	0653	02
(a)	.,	/Fe; ium/Na;		[1] [1]
(b)	(b) alloy is a light material/ has a low density; low mass material need for planes; less fuel needed; alloy is stronger; alloy resists corrosion; (allow does not corrode but reject the word rust) aircraft does not break up in flight;			
(c)	redu	oxide; uction is loss of oxygen / or strong implication; o allow description of electron gain by <u>iron ions / Fe</u>	<sup>3+</sup> )	[2]
	stee	el is stronger; el is less brittle; el is more resistant to corrosion; <i>(allow it does not ru</i>	est)	[max 2]
(d)	$Cl_2$ ;			[1]

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