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## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2010 question paper for the guidance of teachers

## 0653 COMBINED SCIENCE

0653/22

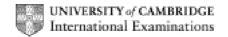
Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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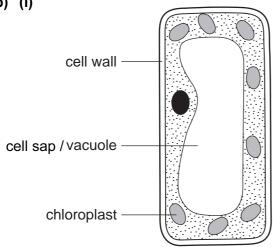
CIE is publishing the mark schemes for the October/November 2010 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



Page 2		)	Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – October/November 2010	0653	22
1	<ul><li>(a) 1. traps (layer of) air;</li><li>2. acts as insulator/reduces conduction/reduces convection</li></ul>			ion ;	[2]
	(b) (i) increased risk of skin cancer/eye damage/sun burn;				[1]
	(ii)		wave ; use ;		[2]
					[Total: 5]
2	(a) (i)		l oxide + carbon → lead + carbon dioxide 5 ; RHS ;		[2]
	<ul> <li>(ii) 1. lead oxide / carbon dioxide; ecf</li> <li>2. compounds contain more than one type of element / atom;</li> <li>3. reference to (different) elements / atoms in compounds being joined / bond</li> </ul>			bonded; [3]	
	(b) (i)	(dc)	power supply / battery / cell;		[1]
	<ul><li>(ii) 1. chlorine;</li><li>2. anode, non-metals form at the anode/chlorine is a non-metal/chloride negative (and anode is positive);</li></ul>		nloride ions are		
					[Total: 8]
3	(a) (i)	tran	spiration / evapotranspiration / diffusion ;		[1]
	(ii)	ston	nata ;		[1]
	(iii)	2. 3.	condensation; water vapour cooled / temperature fell; gas changed to, liquid / water; ref. to particles and (kinetic) energy;		[max 2]

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[max 2]

(ii) palisade (mesophyll);

1

[2]

- (iii) 1. carbon dioxide + water;
  - 2. glucose / carbohydrate / starch / sugar + oxygen;

[Total: 9]

**4** (a) 28(s); [1]

- (b) (i) 1. distance = speed × time; 2. = 10 × 60 × 9 = 5400 m; [2]
  - (ii) 1. work done = force × distance; 2. = 10 000 × 5400 = 54 000 000 J; [2]

[Total: 5]

- **5** (a) idea of restoring full number of chromosomes in the zygote; [1]
  - (b) (i) ovary; [1]
    - (ii) oviduct / Fallopian tube ; [1]
  - (c) 1. produces / contains, (amniotic) fluid;2. protects / supports, embryo;[2]
  - (d) 1. idea that mother's body needs to make substances for both herself and the fetus;
    - 2. iron for haemoglobin;
    - 3. calcium for, bones / teeth; [3]

[Total: 8]

1	Page 4			Mark Scheme: Teachers' version Syllabi			
			IGCSE – October	r/November 2010	0653	3 22	
6	<b>(a)</b> wa	iter conduc	cts electricity;			[1]	
	<b>(b)</b> all	symbols ir	n correct circuit ;;;				
	all	four correc					
			rect for 2 marks ct for 1 mark			[3]	
						1.1	
	(c) (i)	<b>K</b> and <b>L</b>	•			[1]	
	(ii)					roz	
		2. <b>K</b> &	<b>L</b> go off ;			[2]	
	( <b>d</b> ) ad	d one 12 (	to one 8 O (in series	.).		[4]	
	( <b>u</b> ) au	u one 123	$\Omega$ to one 8 $\Omega$ (in series	>),		[1]	
i	(e) B	F C	D E A				
	B D	F;				[0]	
	D	E;				[2]	
						[Total: 10]	
7	(a) (i)	O and S	:			[1]	
-			,			1.1	
	(ii)		element name	protons	neutrons		
			(oxygen)	8	8		
			phosphorus	(15)	(16)		
		ana mar	for each row :	,		[2]	
	one mark for each row ;;					[2]	
	(b) (i)						
	·-/ (·/	copper o	xide / copper carbona	ate / other correct;		[1]	
				ate / other correct ;			
	(ii)	magnesi		ate / other correct ;		[1] [1]	
	(ii)	magnesi (allow al	um ; uminium)	ate / other correct ;			
		magnesi (allow all	um ; uminium) <b>1</b>	ate / other correct ;			
ı	(ii)	magnesi (allow all reaction combust	um ; uminium) <b>1</b> ion / oxidation ;	ate / other correct ;			
	(ii)	magnesi (allow all	um ; uminium) 1 ion / oxidation ;	ate / other correct ;			
1	(ii) (c) (i)	magnesi (allow all reaction combust reaction polymeri	um; uminium)  1 ion / oxidation;  2 sation;			[1]	
1	(ii)	magnesi (allow all reaction combust reaction polymeri	um ; uminium) 1 ion / oxidation ;			[1]	

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**8** (a) (i) 2000 (kg per hectare);

[1]

(ii) any two values with a range between 6 and 7.25;

[1]

- (iii) 1. calcium carbonate is a base;
  - 2. neutralise (acid in the soil);
  - 3. raise pH;
  - 4. above 5.5 / closer to 6.0;

[2 max]

- (b) 1. terracing/walls (qualified);
  - 2. bunds/embankment/ditch;
  - 3. plough along slope (not up and down);
  - 4. keep crop cover (at all times)/plant trees;
  - 5. other valid points;;

[2 max]

(c) (i) insects; allow self-pollination

[1]

- (ii) 1. pollen contains male gamete;
  - 2. fertilisation must take place;
  - 3. male gamete (in pollen grain) fuses with female gamete;
  - 4. seed develops from (fertilised) ovule;

[2 max]

- (iii) 1. biuret test/add biuret reagent/add copper sulfate and KOH solution;
  - 2. purple/lilac/mauve;

[2]

[Total: 11]

9 (a)

	description	charge	range in air	ionising ability
alpha	helium nucleus	positive	5 cm	very strong
beta	electron	negative	50 cm	medium
gamma	wave	none	many kilometres	weak

;;;; [4]

(b) alpha particles have low penetration in air/ will not reach people living in house/ smoke detectors are a long way from people;

[1]

[Total: 5]

Page 6	Mark Scheme: Teachers' version	Syllabus	Paper
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- 10 (a) 1. adding chlorine;
  - 2. kills (harmful) bacteria / microorganisms / germs;
  - filtration;
  - 4. removes solid / insoluble materials / dirt;

4

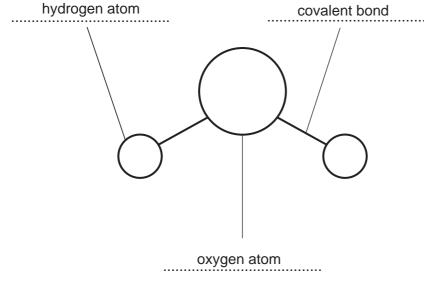
- (b) (i) 1. dissolves in / reacts with rain water;
  - 2. produces acidic solution / acid rain;
  - 3. ref. to sulfurous / sulfuric acid;
  - 4. acidic rain collects in rivers / lakes;
  - 5. reference to harmful effects of acidity, e.g. kills organisms;

max 3

(ii) removal of sulfur compounds from fuel / removal of sulfur dioxide from waste gases / reduce demand for energy / burn less fuel;

max 1

(c)



3

[Total: 11]