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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2010 question paper for the guidance of teachers

0653 COMBINED SCIENCE

0653/23

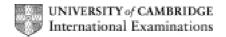
Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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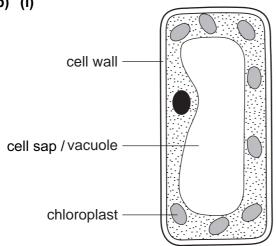
CIE is publishing the mark schemes for the October/November 2010 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



Page 2)	Mark Scheme: Teachers' version	Syllabus	Paper
			IGCSE – October/November 2010	0653	23
1	(a) 1. 2.		s (layer of) air ; as insulator / reduces conduction / reduces convect	ion ;	[2]
	(b) (i) increased risk of skin cancer/eye damage/sun burn;				[1]
	(ii)	(ii) 1. wave; 2. use;			[2]
					[Total: 5]
2	(a) (i) lead oxide + carbon → lead + carbon dioxide LHS; RHS;				[2]
	 (ii) 1. lead oxide / carbon dioxide; ecf 2. compounds contain more than one type of element / atom; 3. reference to (different) elements / atoms in compounds being joined / bon 			bonded; [3]	
	(b) (i)	(dc)	power supply / battery / cell;		[1]
	(ii) 1. chlorine;2. anode, non-metals form at the anode/chlorine is a non-metal/c negative (and anode is positive);		nloride ions are [2]		
					[Total: 8]
3	(a) (i)	trans	spiration / evapotranspiration / diffusion ;		[1]
	(ii)	ston	nata ;		[1]
	(iii)	2. 3.	condensation ; water vapour cooled / temperature fell ; gas changed to, liquid / water ; ref. to particles and (kinetic) energy ;		[max 2]

Page 3	Mark Scheme: Teachers' version	Syllabus	Paper
	IGCSE – October/November 2010	0653	23





[max 2]

(ii) palisade (mesophyll);

1

[2]

- (iii) 1. carbon dioxide + water;
 - 2. glucose / carbohydrate / starch / sugar + oxygen;

[Total: 9]

- (b) (i) 1. distance = speed × time; 2. = 10 × 60 × 9 = 5400 m; [2]
 - (ii) 1. work done = force × distance; 2. = 10 000 × 5400 = 54 000 000 J; [2]

[Total: 5]

- **5** (a) idea of restoring full number of chromosomes in the zygote; [1]
 - (b) (i) ovary; [1]
 - (ii) oviduct / Fallopian tube ; [1]
 - (c) 1. produces / contains, (amniotic) fluid;2. protects / supports, embryo;[2]
 - (d) 1. idea that mother's body needs to make substances for both herself and the fetus;
 - 2. iron for haemoglobin;
 - 3. calcium for, bones / teeth; [3]

[Total: 8]

Pag	Page 4		Mark Scheme: To			labus	Paper
	IGCSE – October/November 2010			0	653	23	
6 (a)	(a) water conducts electricity;						[1]
(b)	(b) all symbols in correct circuit ;;;						
	all f	our correc					
			rect for 2 marks It for 1 mark				[3]
							[-1
(c)	(i)	K and L	•				[1]
((ii)	1. J ligi					ro1
		2. K &	L go off;				[2]
(d)	add	ono 12 () to one 9 () (in corios	.) .			[4]
(u)	auu	OHE 123	Ω to one 8 Ω (in series	·) ,			[1]
(e)	В	F C	D E A				
	В	F;					[0]
	D	Ε;					[2]
							[Total: 10]
7 (a)	(i)	O and S					[1]
		o and o	,				[1]
((ii)		element name	protons	neutrons		
			(oxygen)	8	8		
			phosphorus	(15)	(16)		
							ro1
	one mark for each row ;;					[2]	
(b)	(i)	copper o	xide / copper carbona	te / other correct;			[1]
((ii)	magnesium;			[1]		
`	(,		uminium)				1.1
(c)	(i)	reaction combust	1 ion / oxidation;				
		reaction polymeri					[2]
((ii)	(molecul	es) join together / form	n chains :			[1]
`	· · · · · · · · · · · · · · · · · · ·						
							[Total: 8]

Page 5	Mark Scheme: Teachers' version	Syllabus	Paper
	IGCSE – October/November 2010		23

8 (a) (i) 2000 (kg per hectare);

[1]

(ii) any two values with a range between 6 and 7.25;

[1]

- (iii) 1. calcium carbonate is a base;
 - 2. neutralise (acid in the soil);
 - 3. raise pH;
 - 4. above 5.5 / closer to 6.0;

[2 max]

- (b) 1. terracing/walls (qualified);
 - 2. bunds/embankment/ditch;
 - 3. plough along slope (not up and down);
 - 4. keep crop cover (at all times) / plant trees;
 - 5. other valid points;;

[2 max]

(c) (i) insects; allow self-pollination

[1]

- (ii) 1. pollen contains male gamete;
 - 2. fertilisation must take place;
 - 3. male gamete (in pollen grain) fuses with female gamete;
 - 4. seed develops from (fertilised) ovule;

[2 max]

- (iii) 1. biuret test / add biuret reagent / add copper sulfate and KOH solution;
 - 2. purple/lilac/mauve;

[2]

[Total: 11]

9 (a)

	description	charge	range in air	ionising ability
alpha	helium nucleus	positive	5 cm	very strong
beta	electron	negative	50 cm	medium
gamma	wave	none	many kilometres	weak

;;;; [4]

(b) alpha particles have low penetration in air/ will not reach people living in house/ smoke detectors are a long way from people;

[1]

[Total: 5]

Page 6	Page 6 Mark Scheme: Teachers' version		Paper
	IGCSE – October/November 2010	0653	23

- 10 (a) 1. adding chlorine;
 - 2. kills (harmful) bacteria / microorganisms / germs;
 - filtration;
 - 4. removes solid/insoluble materials/dirt;

4

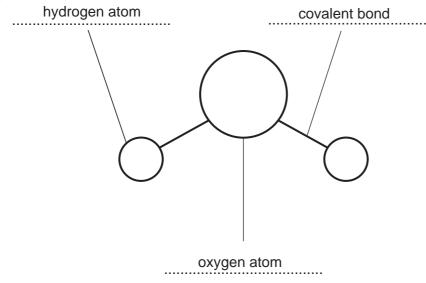
- (b) (i) 1. dissolves in / reacts with rain water;
 - 2. produces acidic solution / acid rain;
 - 3. ref. to sulfurous / sulfuric acid;
 - 4. acidic rain collects in rivers / lakes;
 - 5. reference to harmful effects of acidity, e.g. kills organisms;

max 3

(ii) removal of sulfur compounds from fuel / removal of sulfur dioxide from waste gases / reduce demand for energy / burn less fuel;

max 1

(c)



3

[Total: 11]