MARK SCHEME for the October/November 2012 series

0654 CO-ORDINATED SCIENCES

0654/61

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2012 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.



Page 2			Mark Scheme	Syllabus	Paper		
				IGCSE – October/November 2012	0654	61	
1	(a)	(i)	first	<i>row</i> : 10, 10, 10, 10 ;		[1]	
		(ii) second row: 0, 9, 0, 10 ;					
	(b)	 (b) light not necessary ; water is necessary ; 					
	(c)	to improve reliability/because some seeds might be dead or damaged/to take					
		account of individual variability ;					
	(d)	•	' two o				
			iperat gen/a				
		рН	;			[max 2]	
	(\mathbf{a})	sta	rch	soods :			
	(6)	(e) starch – seeds ; reducing sugar – radicles / roots ;					
	(f)	am	ylase	/carbohydrase/diastase ;		[1]	
						[Total: 10]	
n	(a)	(1)	6 A E				
2	(a)	(1)	64.5 59.2			[2]	
		(ii)	(64.5	5-40=) 24.5 <i>and</i> (59.2-40=) 19.2 (both correct)	;	[1]	
		(iii)	1/70	= 0.014 ;			
				= 0.011 ; alise incorrect d.p. once only)		[2]	
			(1			[-]	
	(b)	(i)		ect plots of 4 or 5 points ;		101	
				ght line drawn ;		[2]	
		(ii)		nd <i>y</i> - distances shown on graph ; correctly calculated (1600 to 1800) ;		[2]	
			-				
	(c)			radient/10 correctly calculated from candidate's g	raph (around 12		
		140	<i>i</i>), uo	not allow impossible masses e.g. negative ;		[1]	
						[Total: 10]	

Page 3	Mark Scheme	Syllabus	Paper	
	IGCSE – October/November 2012	0654	61	
(a) same ma	ass of soil/same volume of water ;		[1]	
(b) (from) bl (to) red ;			[2]	
(c) (i) 4.4 4.9 5.2			[3]	
(ii) 5.6,	5.1, 4.8 (all three, ecf) ;		[1]	
(iii) (5.6	+ 5.1 + 4.8 = 15.5, 15.5/3 =) 5.17 OR 5.2 ;		[1]	
	$3 \times 10/5.2 = 0.05 \text{ (mol/dm}^3) \text{ (ecf)};$ nore d.p.)		[1]	
(e) the (inso	luble) <u>hydroxides</u> (of the metals) are formed/owtte	,	[1]	
			[Total: 10]	

condition of leaves	time/ mins	reading on scale/ cm	distance moved by bubble per minute/cm	average distance moved by bubble per minute/cm
	1	1.6	1.6	1.57
untreated	2	3.3	1.7	OR 1.6
	3	4.7	1.4	

(i)	as in column 3 ;	[1]
(ii)	as in column 4 ;	[1]
(iii)	as in column 5 ;	[1]
(b) (i)	1.2/1.6 × 100 ; = 75 % ; (accept 76 % if 1.57 used)	[2]
(ii)	cover the lower surface with grease (this should stop all transpiration);	

 (ii) cover the lower surface with grease (this should stop all transpiration); (candidates may suggest to repeat the experiment, this time with untreated and then lower surface greased. the mark should be allowed for this)

[1]

	Page 4				yllabus	Paper	
				IGCSE – October/November 2012	0654	61	
	(c) any two of: change in air speed; temperature; humidity; light;						
	(d)	(i)	to p	prevent air bubbles from entering the shoot ;		[1]	
		(ii)		er used in plant for photosynthesis/maintaining ansion/produced by respiration ;	cell turgor/	cell [1] [Total: 10]	
5	(a)		green purple	n ; e/blue ;		[2]	
	(b)	(so	dium)) sulfate ;		[1]	
	(c)) chloride ;) nitrate ;		[2]	
	(d)	(i) (i	(litr	mus is blue at first and then) turns red ; mus is blue at first and then) turns red ; bbles are given off ;		[3]	
	(e)	(i)		um sulfate;		[1]	
	(-)			lid is formed from a solution/insoluble solid forms ;		[1]	
		(")	a 30			[Total: 10]	
6	(a)	(i)	heat light	t ; ; (either order)		[2]	
		(ii)	argo	on OR inert gas ;		[1]	
	(b)	A a	nd V	shown in correct places in the circuit ;		[1]	
	(c)	0.6 12 `				[2]	

Page 5	Mark Scheme	Syllabus	Paper
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(d) (i) 150	0/240 = 0.6(25) A ;		[1]
• •	resistance must be much higher at the higher e her temperature) ;	.m.f. (because of	the [1]
so that	(electrical) energy is wasted/not needed/lost ; energy needs to be generated/fossil fuels need t	o be used (to m	nake
	· y / ,		[max 2]
			[Total: 10]